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Solubility Curves. When we looked at the solubility rules, we classified salts as either soluble or insoluble in water. In reality there is no salt that is 100% insoluble.

Solubility Curves Lab - Saturated Solutions and Precipitates

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Each point on the curve represents a saturated solution, any point below the curve an unsaturated solution and any point above the curve a supersaturated solution. We can use solubility curves to:

SOLUBILITY CURVES - TSFX

DETERMINING A SOLUBILITY CURVE AIM: To obtain the solubility curve for potassium chlorate, KClO₃
BACKGROUND: A solubility curve is determined for a substance so that chemists know how much solute they can add to a solvent in order to obtain a saturated solution. APPARATUS: 4 g pure crystalline potassium

chlorate burette large clean test tube and glass stirring rod thermometer 150 mL distilled ...

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